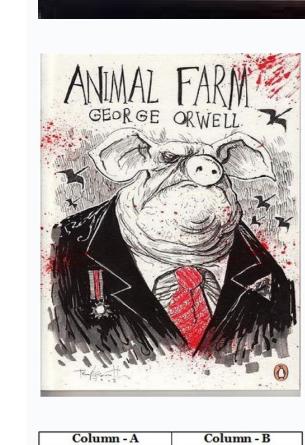
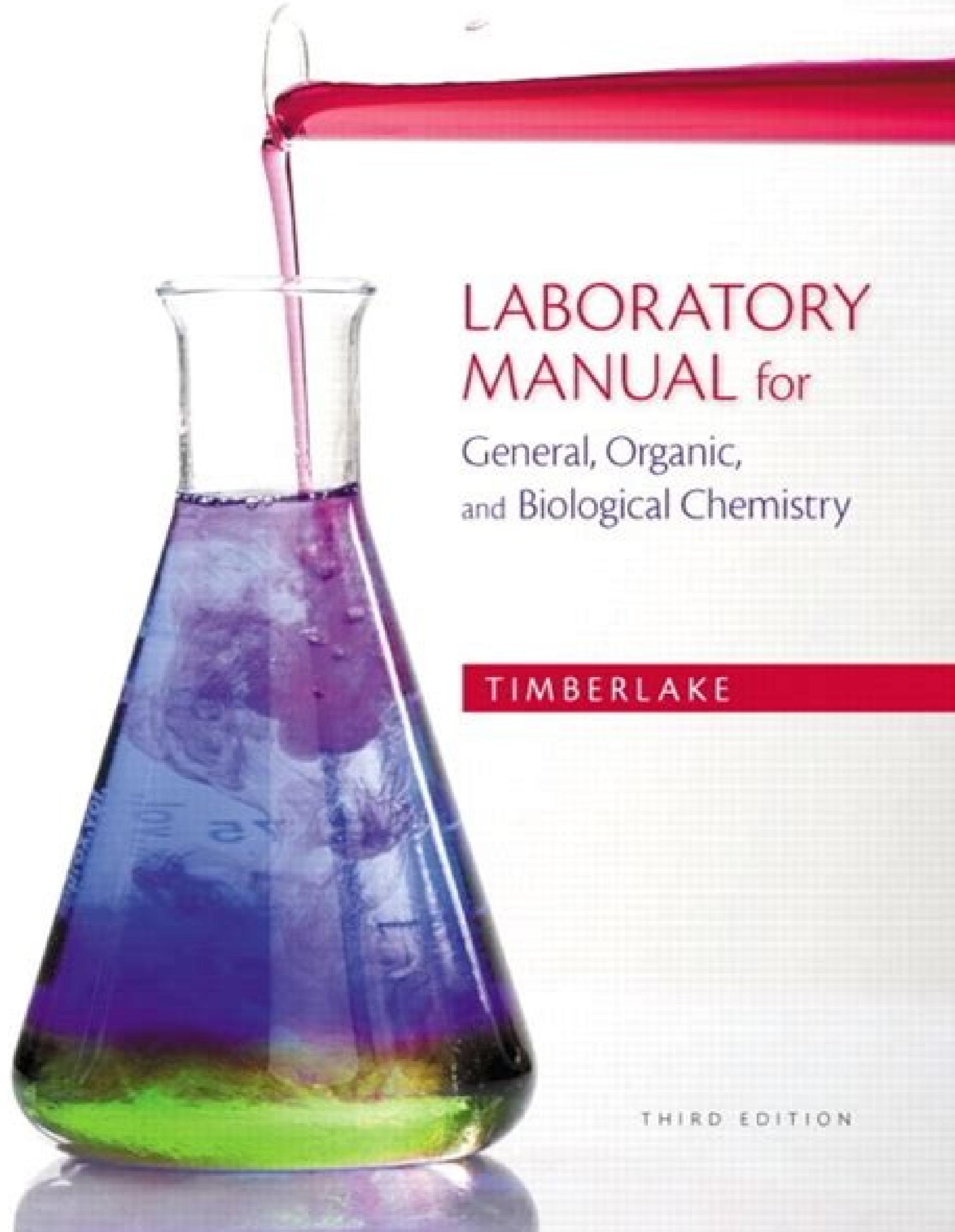
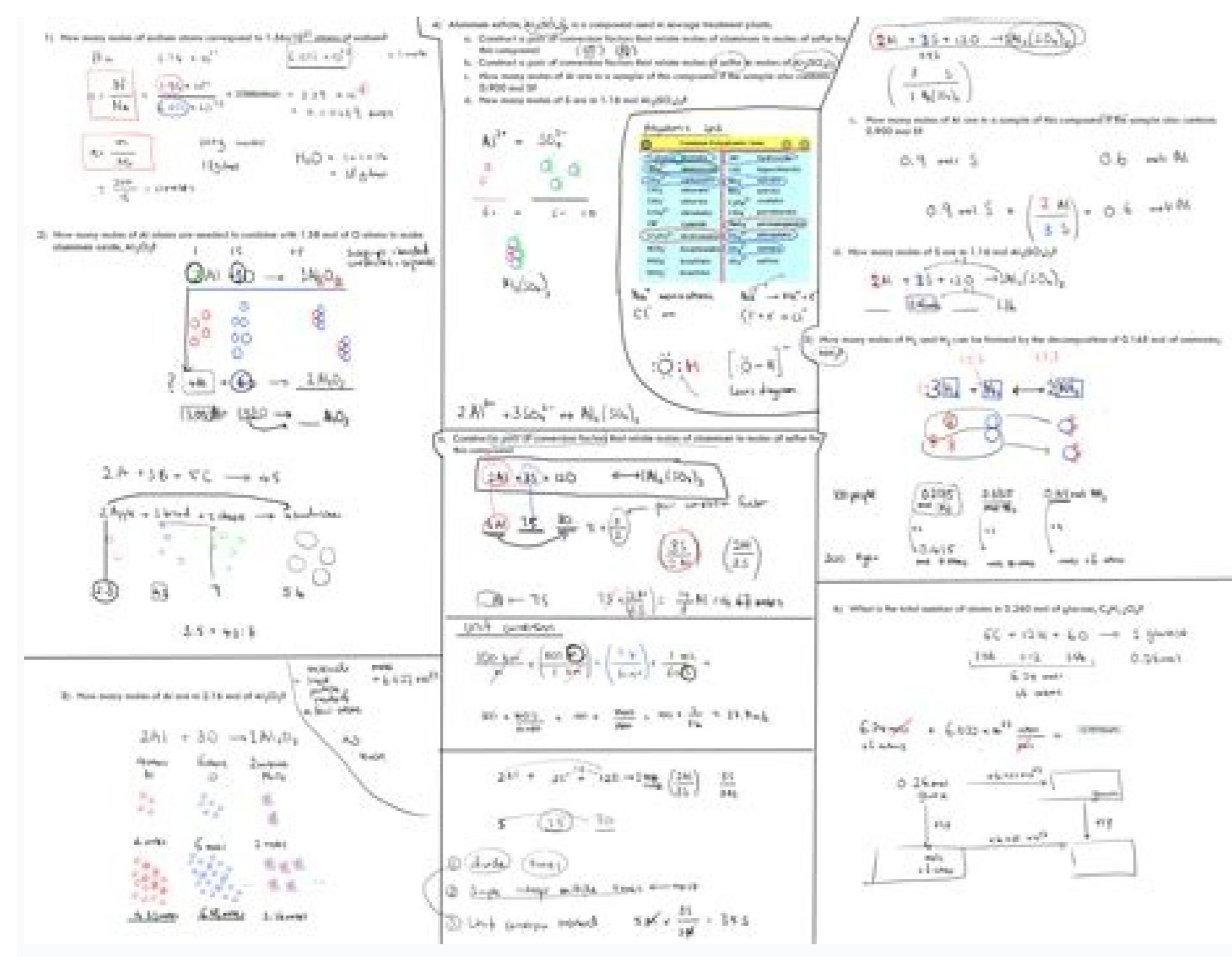


I'm not a robot!





Column A Column B

A) Oil from water. F) Magnetic separation

B) Iron from sand. G) Sedimentation

C) Iron from sand. H) Filtration

D) Iron from sand. I) Separating funnel

E) For salt water. J) Separating funnel

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ISBN 978-0-39312-417-0 Publisher University Science Books ISBN 978-1-89138-960-3 Publisher Cengage Learning ISBN 978-1-28505-137-6 Publisher Pearson ISBN 978-0-32176-452-2 Publisher Pearson ISBN 978-0-32196-746-6 Publisher McGraw-Hill Education ISBN 978-0-07802-151-8 Publisher Pearson ISBN 978-0-32176-414-4 Publisher Brooks/Cole ISBN 978-1-31904-659-5 Publisher Wiley & Sons ISBN 978-1-11801-040-2 Publisher H. Freeman ISBN 978-0-2929-019-1 2022 Oxford University Press All rights reserved. View Mathematics Pure 3 extra online content finderView Mathematics Pure 3 extra online content finderView Mathematics Statistics 3 extra online content finderView Mathematics Decision 1 extra online content finderView Mathematics Mechanics 3 extra online content finderView Mathematics Further Pure 1 extra online content finderView Mathematics Further Pure 2 extra online content finderView Mathematics Further Pure 3 extra online content finder 1. Place a glass of water outside. It will freeze if the temperature is below 0 °C. 3. (a) law (states a consistently observed phenomenon, can be used for prediction); (b) theory (a widely accepted explanation of the behavior of matter); (c) hypothesis (a tentative explanation, can be investigated by experimentation); (d) symbolic, microscopic; (e) symbolic, macroscopic; (f) microscopic. The heat required is determined from macroscopic properties. 9. Liquids can change their shape (flow); solids can't. Gases can undergo large volume changes as pressure changes; liquids do not. Gases flow and change volume; solids do not. 11. The mixture can have a variety of compositions; a pure substance has a definite composition. Both have the same composition from point to point. 13. Molecules of elements contain only one type of atom; molecules of compounds contain two or more types of atoms. They are similar in that both are comprised of two or more atoms chemically bonded together. 15. Answers will vary. Sample answer: Gatorade contains water, sugar, dextrose, citric acid, salt, sodium chloride, monopotassium phosphate, and sucrose acetate isohydrate. 17. (a) element; (b) elements; (c) compound; (d) mixture; (e) compound; (f) compound; (g) mixture 19. In each case, a molecule consists of two or more combined atoms. They differ in that the types of atoms change from one substance to the next. 21. Gasoline (a mixture of compounds), oxygen, and to a lesser extent, nitrogen are consumed. Carbon dioxide and water are the principal products. Carbon monoxide and nitrogen oxides are produced in lesser amounts. 23. (a) Increased as it would have combined with oxygen in the air thus increasing the amount of matter and therefore the mass. (b) 0.9 g 25. (a) 200.0 g; (b) The mass of the container and contents would decrease as carbon dioxide is a gaseous product and would leave the container. (c) 102.3 g 27. (a) physical; (b) chemical; (c) chemical; (d) physical; (e) physical 31. The value of an extensive property depends upon the amount of matter being considered, whereas the value of an intensive property is the same regardless of the amount of matter being considered. 33. Being extensive properties, both mass and volume are directly proportional to the amount of substance under study. Dividing one extensive property by another will in effect "cancel" this dependence on the amount of matter, yielding a ratio that is independent of the amount of matter being considered. 35. Being intensive properties, both mass and volume are directly proportional to the amount of substance under study. Dividing one intensive property by another will in effect "cancel" this dependence on the amount of matter, yielding a ratio that is independent of the amount of matter being considered. 37. (a) kilograms; (b) meters; (c) meters/second; (d) kilograms/cubic meter; (e) kelvin; (f) square meters; (g) cubic meters 39. (a) centi-,  $\times 10^{-2}$ ; (b) deci-,  $\times 10^{-1}$ ; (c) Giga-,  $\times 10^9$ ; (d) kilo-,  $\times 10^3$ ; (e) milli-,  $\times 10^{-3}$ ; (f) nano-,  $\times 10^{-9}$ ; (g) pico-,  $\times 10^{-12}$ ; (h) tera-,  $\times 10^{12}$  41. (a) m = 18.58 g V = 5.7 mL; (b) d = 3.3 g/mL; (c) dioplate (copper cyclosilicate, d = 3.28–3.31 g/mL); malachite (basic copper carbonate, d = 3.25–4.10 g/mL); Paraiba tourmaline (sodium lithium boron silicate with copper, d = 2.82–3.32 g/mL) 43. (a) displaced water volume = 2.8 mL; (b) displaced water mass = 2.8 g; (c) The block mass is 2.76 g, essentially equal to the mass of displaced water (2.8 g) and consistent with Archimedes' principle of buoyancy 45. (a) 7.04  $\times 10^2$ ; (b) 3.344  $\times 10^2$ ; (c) 5.479  $\times 10^2$ ; (d) 2.2086  $\times 10^4$ ; (e) 1.00000  $\times 10^3$ ; (f) 6.51  $\times 10^{-8}$ ; (g) 7.157  $\times 10^{-3}$  47. (a) exact; (b) exact; (c) uncertain; (d) exact; (e) uncertain; (f) uncertain 49. (a) two; (b) three; (c) five; (d) four; (e) six; (f) two; (g) five 51. (a) 0.44; (b) 9.0; (c) 27; (d) 140; (e) 0.44 53. (a) 2.15  $\times 10^5$ ; (b) 4.2  $\times 10^6$ ; (c) 2.08; (d) 0.19; (e) 27,440; (f) 43.0 55. (a) Archer X; (b) Archer W; (c) Archer Y 57. (a) 1.0936 yd<sup>3</sup> m<sup>1</sup> 0.936 yd<sup>1</sup> m<sup>3</sup>; (b) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (c) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (d) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (e) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (f) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (g) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (h) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (i) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (j) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (k) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (l) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (m) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (n) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (o) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (p) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (q) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (r) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (s) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (t) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (u) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (v) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; (w) 1.35 x 10<sup>-3</sup> m<sup>3</sup>; 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